

# FORMATION OF COPPER SULFIDE LAYERS ON POLYPROPYLENE BY CBD METHOD

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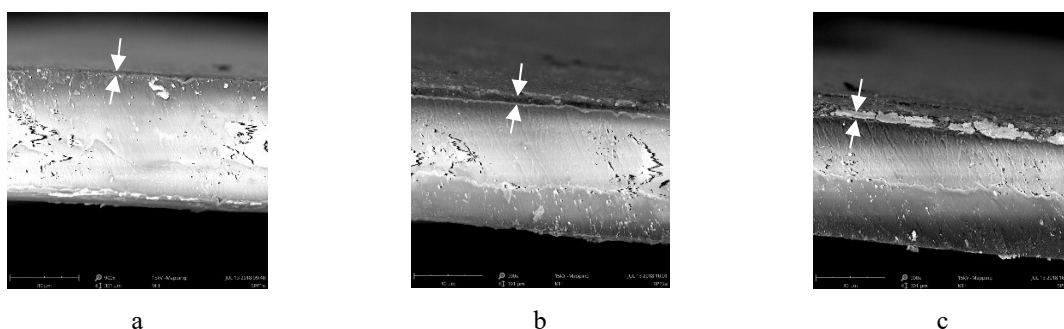
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Nanocrystalline copper sulfide ( $\text{Cu}_x\text{S}$ ), deposited as a thin film on the different substrates, is considered as a promising material for solar energy conversation system, due to their structural and electrical properties. One of the interesting semiconducting layers is copper sulfide due to its metal-like electrical conductivity, chemical-sensing capability, and ideal characteristics for solar energy absorption. Consequently, polymers modified with copper sulfide layers are used as the conductive substrates for deposition of metal and semiconductors; as gas sensors functioning at temperatures tending to room temperature; in thermoelectric applications; in solar cells; in lithium ion batteries.

We have chosen polypropylene (PP) film as a cheap, chemically stable, and flexible substrate. The hydrophobic PP requires an initial surface pre-treatment and weight losses after chemical treatment are recommended to be  $2 \text{ g m}^{-2}$ . This result was reached when PP film was etched for 25 min at  $90^\circ\text{C}$  with oxidizing solution ( $\text{H}_2\text{SO}_4/\text{H}_3\text{PO}_4$  (1:1), saturated with  $\text{CrO}_3$ ).

The  $\text{Cu}_x\text{S}$  thin films deposit was carried out at room temperature by using the following procedure: 0.05 M  $\text{CuCl}_2$  and 0.05 M  $\text{Na}_2\text{S}_2\text{O}_3$  were mixed, the pH of the resultant solution was adjusted to 3. The pre-treated PP samples were immersed vertically along the wall of the reactor and were left undisturbed for deposition of  $\text{Cu}_x\text{S}$  films for 16 h at  $20^\circ\text{C}$ . At the end of the deposition time, the samples were taken out, and then rinsing of the substrate with distilled water for 30 s removed the desorbed ions and dried in a desiccator for 8 h. The deposition process was carried out by repeating such deposition cycles 3 times. Formed samples were annealed at  $80^\circ\text{C}$  for 30 min. Electrical resistivity was measured, and SEM/EDS investigation were performed after each formation cycle and after annealing of the samples.

The room temperature electrical resistivity of the formed layer after 1st deposition cycle was about  $10 \text{ k}\Omega/\square$ , after 2nd –  $180 \Omega/\square$ , and after 3rd –  $130 \Omega/\square$ . After annealing electrical resistivity decreased to  $7 \text{ k}\Omega/\square$ ,  $170 \Omega/\square$  and  $110 \Omega/\square$  accordingly.



**Fig. 1.** Cross section of PP/ $\text{Cu}_x\text{S}$  composite after: a) 1st deposition cycle, b) 2nd deposition cycle; c) 3rd deposition cycle

Analysis of SEM images of PP/ $\text{Cu}_x\text{S}$  films showed that after 3 deposition cycles, the PP substrate was completely covered by  $\text{Cu}_x\text{S}$  particles. Cross sectional image (Fig. 1) demonstrated growing of  $\text{Cu}_x\text{S}$  layer with increasing of the cycle number.