

FORMATION OF COPPER SULFIDE LAYERS ON POLYPROPYLENE BY CBD METHOD

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Nanocrystalline copper sulfide (Cu_xS), deposited as a thin film on the different substrates, is considered as a promising material for solar energy conversion system, due to their structural and electrical properties. One of the interesting semiconducting layers is copper sulfide due to its metal-like electrical conductivity, chemical-sensing capability, and ideal characteristics for solar energy absorption. Consequently, polymers modified with copper sulfide layers are used as the conductive substrates for deposition of metal and semiconductors; as gas sensors functioning at temperatures tending to room temperature; in thermoelectric applications; in solar cells; in lithium ion batteries.

We have chosen polypropylene (PP) film as a cheap, chemically stable, and flexible substrate. The hydrophobic PP requires an initial surface pre-treatment and weight losses after chemical treatment are recommended to be 2 g m^{-2} . This result was reached when PP film was etched for 25 min at 90°C with oxidizing solution ($\text{H}_2\text{SO}_4/\text{H}_3\text{PO}_4$ (1:1), saturated with CrO_3).

The Cu_xS thin films deposit was carried out at room temperature by using the following procedure: 0.05 M CuCl_2 and 0.05 M $\text{Na}_2\text{S}_2\text{O}_3$ were mixed, the pH of the resultant solution was adjusted to 3. The pre-treated PP samples were immersed vertically along the wall of the reactor and were left undisturbed for deposition of Cu_xS films for 16 h at 20°C . At the end of the deposition time, the samples were taken out, and then rinsing of the substrate with distilled water for 30 s removed the desorbed ions and dried in a desiccator for 8 h. The deposition process was carried out by repeating such deposition cycles 3 times. Formed samples were annealed at 80°C for 30 min. Electrical resistivity was measured, and SEM/EDS investigation were performed after each formation cycle and after annealing of the samples.

The room temperature electrical resistivity of the formed layer after 1st deposition cycle was about $10 \text{ k}\Omega/\square$, after 2nd – $180 \Omega/\square$, and after 3rd – $130 \Omega/\square$. After annealing electrical resistivity decreased to $7 \text{ k}\Omega/\square$, $170 \Omega/\square$ and $110 \Omega/\square$ accordingly.

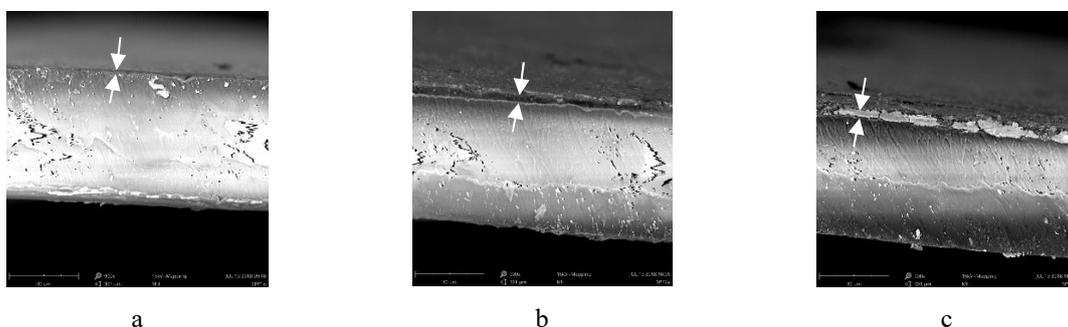


Fig. 1. Cross section of PP/ Cu_xS composite after: a) 1st deposition cycle, b) 2nd deposition cycle; c) 3rd deposition cycle

Analysis of SEM images of PP/ Cu_xS films showed that after 3 deposition cycles, the PP substrate was completely covered by Cu_xS particles. Cross sectional image (Fig. 1) demonstrated growing of Cu_xS layer with increasing of the cycle number.